2.2. DYNAMIC MODELS—LUMPED PARAMETER SYSTEMS

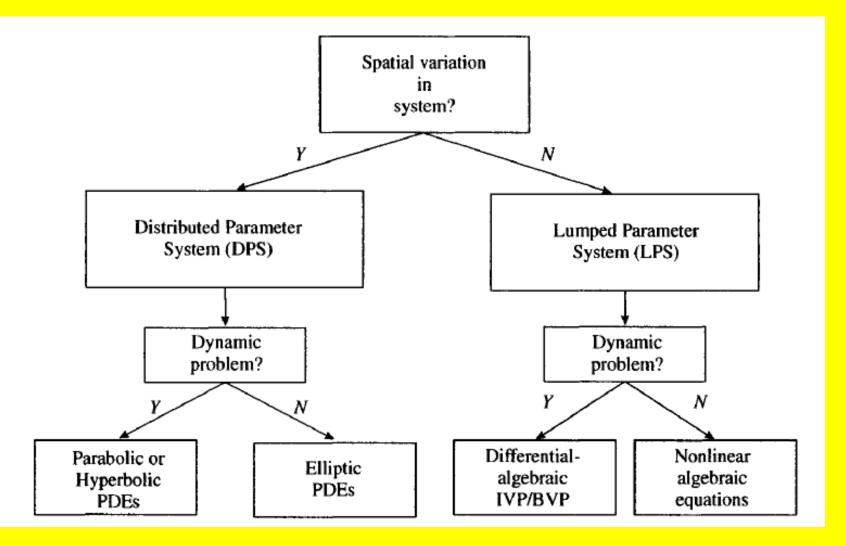
- This section focused on the underlying modelling principles for **lumped parameter systems (LPSs)** and the subsequent analysis of those models.
- Here, we are concerned about the time varying behaviour of systems which are considered to have states which are homogeneous within the balance volume V. Hence, the concept of "lumping" the scalar field into a representative single state value.
- Sometimes the term "well mixed" is applied to such systems where the spatial variation in the scalar field, which describes the state of interest, is uniform.
- In some cases, we may also consider the stationary behaviour of such systems which leads to steady state model descriptions.

- Lumped parameter dynamic models, or compartmental models are widely used **for control and diagnostic purposes**.
- They are frequently used as the basis for engineering design, startup and shutdown studies as well as assessing safe operating procedures.

Dynamic Models

- In terms of dynamic models we have two clearly identifiable classes:
 - distributed parameter dynamic models,
 - lumped parameter dynamic models.
- In the above classes, we identify the distributed parameter dynamic models with various forms of PDEs, principally parabolic partial differential equations (PPDEs).

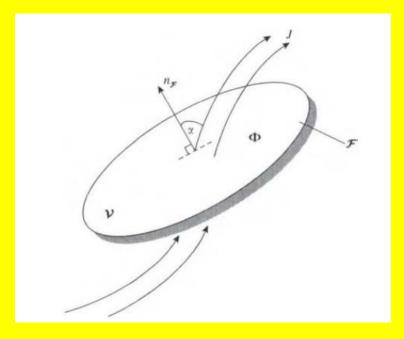
- The lumped parameter dynamic models result in systems of ODEs often coupled with many nonlinear and linear algebraic constraints.
- The total system is referred to as a differential-algebraic equation (DAE) set. The equations need to have a specified set of consistent initial conditions for all states.
- This can be a challenging problem due to the effects of the nonlinear constraints which can impose extra conditions on the choice of initial values.



Classification of Models

PRINCIPLE OF CONSERVATION

- As a result of the first law of thermodynamics, energy is conserved within a system, although it may change its form.
- Also, both mass and momentum in the system will be conserved quantities in any space. In most process systems, we deal with open systems where mass, energy and momentum can flow across the boundary surface.
- As such, we can consider a space with volume V and boundary surface *T* as shown in Fig.



General balance volume V

$$\begin{cases} net change \\ of quantity in time \end{cases} = \begin{cases} flow in \\ through boundary \end{cases} - \begin{cases} flow out \\ through boundary \end{cases} + \begin{cases} net \\ generation \end{cases} - \begin{cases} net \\ consumption \end{cases}.$$

Lumped Conservation Balances

- Lumped parameter models do not incorporate the spatial variation of states within the balance volume; therefore, the scalar field of the intensive quantities (most often concentrations or temperature) is a function of time only.
- What this means is that the application of the general conservation equations leads to models which are represented by ODEs in time.
- Moreover, the closed boundary surface encapsulating the balance volume is also **homogeneous in space**.

Total Mass Balance

• The general expression for a *total mass balance can be* written in word form as:

$$\begin{cases} \text{rate of accumulation} \\ \text{of mass} \end{cases} = \begin{cases} \text{mass flow} \\ \text{in} \end{cases} - \begin{cases} \text{mass flow} \\ \text{out} \end{cases}$$

 or in the case of a lumped parameter system, the equation form for *p input streams* and *q output streams* is

$$\frac{\mathrm{d}M}{\mathrm{d}t} = \sum_{j=1}^p F_j - \sum_{k=1}^q F_k$$

Component Mass Balances

 $\begin{cases} \text{rate of mass accumulation} \\ \text{of component } i \end{cases} = \begin{cases} \text{mass flow in of} \\ \text{component } i \end{cases} - \begin{cases} \text{mass flow out of} \\ \text{component } i \end{cases} \\ + \begin{cases} \text{rate of formation or consumption} \\ \text{of component } i \end{cases} \end{cases}$

- Where the last term accounts for the creation or disappearance of component i via chemical reaction.
- In the case of an LPS, we have the general equation in the form

$$\frac{\mathrm{d}m_i}{\mathrm{d}t} = \sum_{j=1}^p f_{i,j} - \sum_{k=1}^q f_{i,k} + g_i, \quad i = 1, \dots, n,$$

 Where m_i is the mass holdup of component i within the balance volume V, j is the stream number and gi is the mass rate of generation or consumption of species i in the balance volume due to reaction.

- We can also write the general mass balance in molar terms n_i by introducing the molar flowrates instead of the mass flowrates of species i in stream *j*.
- The general molar balance is written as

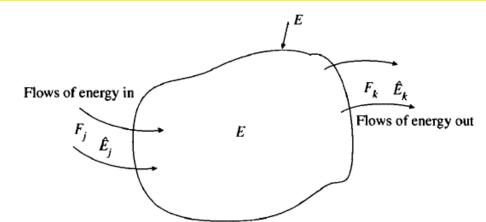
$$\frac{\mathrm{d}n_i}{\mathrm{d}t} = \sum_{j=1}^p \widetilde{f}_{i,j} - \sum_{k=1}^q \widetilde{f}_{i,k} + \widetilde{g}_i, \quad i = 1, \dots, n.$$

Total Energy Balance

 The general conservation balance for total energy over the balance volume V with surface is given by

 $\begin{cases} \text{rate of change} \\ \text{of total energy} \end{cases} = \begin{cases} \text{flow of energy} \\ \text{into the system} \end{cases} - \begin{cases} \text{flow of energy} \\ \text{out of system} \end{cases}$

- The total energy E [J] of the system comprises three principal components in process systems:
 - internal energy U
 - kinetic energy K_E
 - potential energy P_E
- Hence, we can write



$$E = U + K_E + P_E$$

The conservation balance for energy over the balance volume
 V can also be written as

$$\frac{\mathrm{d}E}{\mathrm{d}t} = \sum_{i=1}^{p} F_{j}\hat{E}_{j} - \sum_{i=1}^{q} F_{k}\hat{E}_{k} + Q + W$$

- Where Q is the heat transfer to the surrounding and W is the work done
- Using the thermodynamic relationship for enthalpy H given by H = U + PV, it is possible to write the general energy balance using mass specific enthalpy H [J/kg] as

$$\frac{\mathrm{d}E}{\mathrm{d}t} = \sum_{j=1}^{p} F_j (\hat{H} + \hat{K}_E + \hat{P}_E)_j - \sum_{k=1}^{q} F_k (\hat{H} + \hat{K}_E + \hat{P}_E)_k + Q + \hat{W}$$

Simplifications and Modifications of the General Energy Balance

- Assumption 1 : In many cases, the kinetic and potential energy components can be neglected
- This is a common representation in **chemical process systems** where internal energy content often **dominates the total energy** content of the system. The equation will be simplified to :

$$\frac{\mathrm{d}U}{\mathrm{d}t} = \sum_{j=1}^{p} F_j \hat{H}_j - \sum_{k=1}^{q} F_k \hat{H}_k + Q + \hat{W}$$

 Computation of the specific enthalpies for all inlet and outlet streams is usually done using thermodynamic prediction packages which can take into account fluid phase non-idealities.

- Assumption 2 We normally do not deal directly with the internal energy *U*, in the general energy balance but prefer to use alternate properties.
- Using the definition of enthalpy we can write the above equation as

$$\frac{\mathrm{d}U}{\mathrm{d}t} = \frac{\mathrm{d}(H - PV)}{\mathrm{d}t} = \sum_{j=1}^{p} F_{j}\hat{H}_{j} - \sum_{k=1}^{q} F_{k}\hat{H}_{k} + Q + \hat{W},$$
$$\frac{\mathrm{d}H}{\mathrm{d}t} - \frac{\mathrm{d}(PV)}{\mathrm{d}t} = \sum_{j=1}^{p} F_{j}\hat{H}_{j} - \sum_{k=1}^{q} F_{k}\hat{H}_{k} + Q + \hat{W}.$$

• If P and V are constant, then we can write,

$$\frac{\mathrm{d}H}{\mathrm{d}t} = \sum_{j=1}^{p} F_j \hat{H}_j - \sum_{k=1}^{q} F_k \hat{H}_k + Q + \hat{W}$$

- Assumption 3 In the above equation, the specific enthalpy of the balance volume and the specific enthalpy at the outlet are not necessarily equal.
- In the case where pressure variations within the balance volume are small, such as in liquid systems or where enthalpy variations due to pressure are small, we can assume that the specific enthalpies are equal and hence:

$$\hat{H}=\hat{H}_k\quad k=1,\ldots,q.$$

• We can then write the energy balance as

$$\frac{\mathrm{d}H}{\mathrm{d}t} = \sum_{j=1}^p F_j \hat{H}_j - \sum_{k=1}^q F_k \hat{H} + Q + \hat{W}.$$

This is the most common form of the energy balance for a liquid system

- Assumption 4 We know that in the above Eq. , the enthalpies are evaluated at the temperature conditions of the feeds (Tj) and also at the system temperature (T).
- By making certain assumptions about the enthalpy representation, we can make further simplifications. In particular, we note that the enthalpy of the feed Hj can be written in terms of the system temperature T.

$$\hat{H}_j(T_j) = \hat{H}_j(T) + \int_T^{T_j} c_{p_j}(T) \,\mathrm{d}T$$

• If we assume that *Cp is a constant, then*

$$\hat{H}_j(T_j) = \hat{H}_j(T) + c_{p_j}(T_j - T)$$

 Hence, we can write our modified energy balance as

$$\frac{\mathrm{d}H}{\mathrm{d}t} = \sum_{j=1}^{p} F_j \left[\hat{H}_j(T) + c_{p_j}(T_j - T) \right] - \sum_{k=1}^{q} F_k \hat{H}(T) + Q + \hat{W}$$

- Assumption 5: It has already been mentioned that when considering reacting systems no explicit appearance of the heat of reaction is seen in the general energy balance.
- This is because the energy gain or loss is seen in the value of the outlet enthalpy evaluated at the system temperature *T*.
- We can now develop the energy balance in a way which makes the reaction term explicit in the energy balance.

$$V\rho c_p \frac{\mathrm{d}T}{\mathrm{d}t} = \sum_{j=1}^p F_j c_{p_j} (T_j - T) + rV(-\Delta H_\mathrm{R}) + Q + \hat{W}$$

• This is the most common form of energy balance for reacting systems

- The first term on the right-hand side represents the energy needed to adjust all feeds to the reactor conditions.
- The second represents the energy generation or consumption at the reactor temperature.
- The last two terms are the relevant heat and work terms

CONSERVATION BALANCES FOR MOMENTUM

- In many systems it is also important to consider the conservation of momentum.
- This is particularly the case in mechanical systems and in flow systems where various forces act.
- These can include pressure forces, viscous forces, shear forces and gravitational forces. Momentum is the product of mass and velocity. We can thus write the general form of the balance applied to a similar balance volume as

$$\begin{cases} \text{rate of change} \\ \text{of momentum} \end{cases} = \begin{cases} \text{rate of momentum} \\ \text{into system} \end{cases} - \begin{cases} \text{rate of momentum} \\ \text{out of system} \end{cases} \\ + \begin{cases} \text{rate of momentum} \\ \text{generation} \end{cases} \end{cases}.$$

- The last term is the summation of all the forces acting on the system. In considering momentum, it is important to consider all components of the forces acting on the system under study.
- This means that the problem is basically a 3D problem. In reality we often simplify this to a I D problem. This alternative expression of the momentum balance is given by:

$$\begin{cases} \text{rate of change} \\ \text{of momentum} \end{cases} = \begin{cases} \text{rate of momentum} \\ \text{into system} \end{cases} - \begin{cases} \text{rate of momentum} \\ \text{out of system} \end{cases} \\ + \begin{cases} \text{sum of all forces} \\ \text{on the system} \end{cases}. \end{cases}$$

• We can write the general momentum balance equation as

$$\frac{\mathrm{d}\mathcal{M}}{\mathrm{d}t} = \mathcal{M}^{(\mathrm{i})} - \mathcal{M}^{(\mathrm{o})} + \sum_{k=1}^{p} \mathcal{F}_{k}$$

- Momentum balances in models for lumped parameter systems appear most often in equations relating convective flows to forces generated by pressure, viscous and gravity gradients.
- These are typically expressed by some form of the general **Bernoulli equation** which incorporates various simplifications.

THE SET OF CONSERVATION BALANCES FOR LUMPED SYSTEMS

- The process model of an LPS consists of a set of conservation balance equations that are ODEs equipped with suitable constitutive equations.
- The balance equations are usually coupled through reaction rate and transfer terms.

STEADY-STATE LUMPED PARAMETER SYSTEMS

- In some circumstances we might be interested only in the steady state of the process.
- The general mass and energy balances can be modified to give the steady-state balances by simply setting all derivative (time varying) terms to zero.
- Hence, we arrive at the equivalent steady-state mass, component mass and total energy balances.
- Steady-state total mass balance

$$0 = \sum_{j=1}^p F_j - \sum_{k=1}^q F_k$$

• Steady-state component mass balance

$$0 = \sum_{j=1}^{p} f_{i,j} - \sum_{k=1}^{q} f_{i,k} + g_i, \quad i = 1, ..., n;$$

• Steady-State energy balance

$$0 = \sum_{j=1}^{p} F_j (\hat{U} + P_1 \hat{V}_1 + \hat{K}_E + \hat{P}_E)_j$$
$$- \sum_{k=1}^{q} F_k (\hat{U} + P_2 \hat{V}_2 + \hat{K}_E + \hat{P}_E)_k + Q + \hat{W}$$

- These equations are typically solved using some form of iterative numerical solver such as Newton's method.
- Steady-state balances form the basis for the substantial number of process flowsheeting programs which are routinely used in the process industries. These include ASPEN PLUS, HYSIM and PRO II.

ANALYSIS OF LUMPED PARAMETER MODELS

Degrees of Freedom Analysis

- In the same way that algebraic equations require a degree of freedom analysis to ensure they are properly posed and solvable, dynamic models also require a similar analysis.
- The basis concept of DOF analysis is to determine the difference between the number of variables (unknowns) in a given problem, and the number of equations that describe a mathematical representation of the problem.
- Thus,

$$N_{\rm DF} = N_{\rm u} - N_{\rm e}$$

- Where N_{DF} is the number of DOF, N_u the number of independent variables (unknowns) and N_e the number of independent equations.
- There are three possible values for NDF to take:
- (a) **NDF = 0** This implies that the number of independent unknowns and independent equations is the same.
 - A unique solution exists.

(b) N_{DF} > 0

- This implies that the number of independent variables is greater than the number of independent equations.

The problem is underspecified and a solution is possible only if some of the independent variables are "fixed" (i.e. held constant) by some external considerations in order that N_{DF} be reduced to zero. Some thought must be given to which system variables are chosen as fixed. In the case of optimization these DOF will be adjusted to give a "best" solution to the problem.

(c) **N**_{DF} < **0**

- This implies that the number of variables is less than the number of equations.

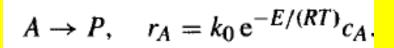
The problem is **overspecified**, meaning that there are **less variables than equations**. If this occurs it is necessary to check and make sure that you have included all relevant variables.

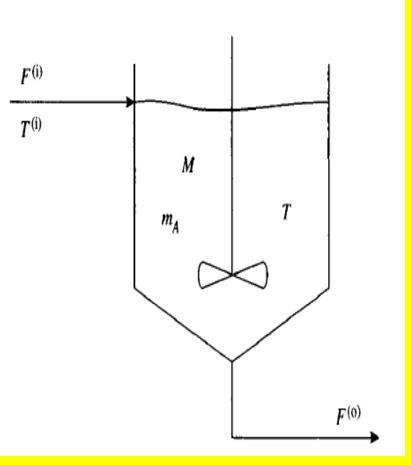
• EXAMPLE (Lumped parameter modelling of a CSTR).

A CSTR is shown in Fig. below with reactant volume V, component mass holdup M_A for component A, feed flowrate Fi [m³/s] at temperature Tⁱ, Feed concentration of component A is c_Aⁱ. Outlet flowrate F⁰ is in units [m³/s].

Assumptions

- Al. perfect mixing implying no spatial variations,
- A2. incompressible fluid phase,
- A3, constant physical properties,
- A4. all flows and properties given in mole units,
- A 5 . equal molar densities,
- A6. reactions and reaction rates given by





Model equations

 The following state the overall mass balance, the component balance for A and the total energy balance.

accumulation-rate = in-flow - out-flow + source,

$$\frac{d(\rho V)}{dt} = F^{(i)}\rho^{(i)} - F^{(o)}\rho,$$

$$\frac{d(m_A)}{dt} = f_A^{(i)} - f_A^{(o)} - Vr_A,$$

$$V\rho c_P \frac{d(T)}{dt} = F^{(i)}\bar{H}^{(i)}\rho - F^{(i)}\bar{H}\rho - Vr_A\Delta H_R.$$

- A set of constitutive equations accompanies the conservation balances. These include:
- The model equations above have to be solved for the following set of state variables:

$$\underline{x} = \begin{bmatrix} V & m_A & T \end{bmatrix}^{\mathrm{T}}$$

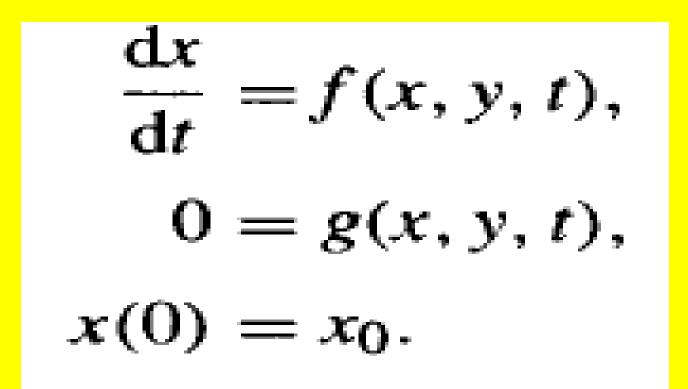
$$f_A^{(i)} = F^{(i)}c_A^{(i)}, \quad f_A^{(0)} = F^{(0)}c_A, \quad M = V\rho, \quad m_A = c_A V,$$

$$r_A = k_0 e^{-E/(RT)}c_A, \quad H = Mc_P T, \quad \bar{H} = \frac{H}{M}, \quad \bar{H}^{(i)} = c_P^{(i)}T^{(i)}.$$

Note that a set of *initial conditions is needed for the solution*. *Initial conditions:*

$$V(0) = V_0; \quad m_A(0) = m_{A_0}; \quad T(0) = T_0.$$

 As shown in the example above, the general form of the lumped parameter model equations is an initial value problem for a set of ODEs with algebraic constraints and initial conditions X(0). This is called a DAE-IVP problem:



High-Index Differential-Algebraic Equations

- The index is the minimum number of differentiations with respect to time that the algebraic system of equations has to undergo to convert the system into a set of ODEs,
- The index of a pure ODE system is zero by definition. If the index of a DAE is one (1), then the initial values of the differential variables can be selected arbitrarily, and easily solved by conventional methods such as Runge-Kutta and Backward Differentiation methods.
- If, however, the index is higher than 1, special care should be taken in assigning the initial values of the variables, since some "hidden" constraints lie behind the problem specifications.

• The requirement of index-1 for a DAE set is equivalent to the requirement that the algebraic equation set should have Jacobian of full rank with respect to the algebraic variables.

• That is

$$g_z = \left(\frac{\partial g}{\partial z}\right)$$

must be non-singular.

EXAMPLE (A linear DAE system). Consider a simple linear DAE system given by

$$\dot{x}_1 = x_1 + x_2 + z_1,$$
 (f₁)
 $\dot{x}_2 = x_1 - x_2 - z_1,$ (f₂)
 $0 = x_1 + 2x_2 - z_1;$ (g₁)

 let us investigate the index of this system. To do so, we differentiate the algebraic constraint gi to get

$$0 = \dot{x}_1 + 2\dot{x}_2 - \dot{z}_1$$
$$\dot{z}_1 = \dot{x}_1 + 2\dot{x}_2.$$

• Substitute for x'₁ and X'₂ from f1 and f2 to get

$$\dot{z}_1 = 3x_1 - x_2 - z_1$$

- Hence, this algebraic constraint has been converted to an ODE after 1 differentiation. This system is INDEX = 1
- As an alternative, consider a change in the algebraic constraint g₁ to

$$0=x_1+2x_2.$$

• Differentiate this for the first time to get

$$0 = \dot{x}_1 + 2\dot{x}_2$$

• and substitute from (f1) and (f2) to get

$$0 = 3x_1 - x_2 - z_1$$

• Clearly, this first differentiation has not produced a differential equation in zi, hence we differentiate once more to get

$$0 = 3\dot{x}_1 - \dot{x}_2 - \dot{z}_1,$$

$$\dot{z}_1 = 3\dot{x}_1 - \dot{x}_2$$

and
$$\dot{z}_1 = 2x_1 + 4x_2 + 4z_1$$

• This result shows that the DAE set is **INDEX = 2**

 It might be asked why the index is of importance in DAE systems. It is not an issue for pure ODE systems but when the INDEX > 1 the numerical techniques which are used to solve such problems fail to control the solution error and can fail completely.

Factors Leading to High-Index Problems

- It has been seen that inappropriate specifications lead to problems with high index.
- There are at least three main reasons why high-index problems arise. These include:
 - (i) Choice of specified (design) variables,
 - (ii) The use of forcing functions on the system,
 - (iii) Modelling issues.
- It must be said that in all the above situations there may be inappropriate cases which lead to a high-index problem. Other situations are valid and truly lead to high-index problems.
- However, numerical routines are generally incapable of handling these high index situations. We prefer to model in such a way that we obtain an index-one (1) problem.

STABILITY OF THE MATHEMATICAL PROBLEM

- The propagation of errors is not only dependent on the type of method used but is influenced dramatically by the behaviour of the problem notably by the integral curves, which represent the family of solutions to the problem.
- This is clearly dependent on the **individual problem**.

- EXAMPLE (Stability of a simple linear ordinary differential system). Before addressing the general nonlinear ODE system, let us look at a simple 2 variable problem to illustrate some basic characteristics
- Consider the problem

$$\frac{dy}{dt} = y' = Ay + \phi,$$

$$\begin{pmatrix} y'_1 \\ y'_2 \end{pmatrix} = \begin{pmatrix} -2000 & 999.75 \\ 1 & -1 \end{pmatrix} \begin{pmatrix} y_1 \\ y_2 \end{pmatrix} + \begin{pmatrix} 1000.25 \\ 0 \end{pmatrix},$$

where $y_1(0) = 0$; $y_2(0) = -2$

• The *exact solution is given by*

$$y_1(t) = -1.499e^{-0.5t} + 0.499e^{-20005t} + 1,$$

$$y_2(t) = -2.999e^{-0.5t} - 0.0025e^{-20005t} + 1.$$

- Note that the first terms in each equation represent the slow transients in the solution whilst the second terms are the fast transients (see the exponential terms). Finally, the constant terms represent the steady state values. The slow transients determine just how long it takes to reach steady state.
- We can note that the fast transient is over at *t* = 0.002 whilst the slow transient is over at *t* = 10.
- If we solved this with a classical Runge-Kutta method, we would need about 7000 steps to reach steady state. Even though the fast transient has died out quickly, the **eigenvalue** associated with this component still controls the steplength of the method when the method has a finite limit on the solution errors.

 Now let us consider the general nonlinear set of ODEs given by

$$y' = f(t, y)$$
 $y(a) = y_0,$ $t \in (a, b).$

 The behaviour of the solution to the problem near a particular solution g(t) can be qualitatively assessed by the linearized variational equations given by:

$$y' = \mathbf{J}(t, g(t))[y - g(t)] + f(t, g(t)),$$

where $\mathbf{J} = \partial f_i / \partial y_j$ is evaluated at (t, g(t)).

- Since the local behaviour is being considered, the Jacobian could be replaced by a constant matrix A provided the variation of J in an interval of t is small.
- Assuming that the matrix A has distinct eigenvalues, λ_i, i= , 1, 2 , . . . , n and that the eigenvectors are V_i i = 1, 2 , . . . , n the general solution of the variational equation has the form

$$y(t) = \sum_{i=1}^{n} c_i e^{\lambda_i t} v_i + g(t).$$

 There are three important cases related to the eigenvalues, which illustrate the three major classes of problems to be encountered.

Unstable Case

- Here, some of the eigenvalues are positive and large, hence the solution curves spread out. A very difficult problem for any ODE method. This is inherent instability in the mathematical problem.
- Consider the solution of the following ordinary differential equation (ODE-IVP):

$$y'=y-t, \quad y(0)=1.$$

- This is an inherently unstable problem with a positive eigenvalue = 1. It is clear that as this problem is integrated numerically, the solution will continue to grow without bound as time heads for infinity.
- Some process engineering problems have this type of characteristic.
- Some catalytic reactor problems can exhibit thermal runaway which leads to an unstable situation when a critical temperature in the reactor is reached.
- Certain processes which have a control system installed can also exhibit instability due to unsatisfactory controller tuning or design.

Stable Case

- Here the eigenvaues have negative real parts and are small in magnitude and hence the solution curves are roughly parallel to g(t).
- These are reasonably easy problems to solve, using conventional explicit techniques like Euler or Runge-Kutta methods. Stable problems are also common in process engineering.

Ultra Stable Case

- Here, some eigenvalues are large and negative (there are others that are small and negative) and the solution curves quickly converge to g(t). This behaviour is good for propagation of error in the ODE but not for a numerical method. This class of problems is called "stiff'.
- When inappropriate numerical methods such as Euler's method is applied to an ultra stable problem then there is bound to be difficulties.
- It should be noted that stiffness is a property of the mathematical problem not the numerical method.

Consider the problem given by

$$\frac{dy}{dt} = Ay, \quad y(0) = [2, 1]^{T}, \quad t \in [0, 10],$$

where the matrix A is given by

$$\mathbf{A} = \begin{pmatrix} -500.5 & 499.5\\ 499.5 & -500.5 \end{pmatrix}$$

The exact solution to this linear ODE is

$$y_1 = 1.5e^{-t} + 0.5e^{-1000t},$$

 $y_2 = 1.5e^{-t} - 0.5e^{-1000t}.$

• Eigen values of the Jacobian of A are

$$\lambda_1 = -1000; \lambda_2 = -1$$

 Hence, the problem has the initial transient followed by the integration of the slower transient as seen in the analytic solutions.